

# Implementing bio-inspired nickel-based nanocatalysts for hydrogen uptake in a PEMFC

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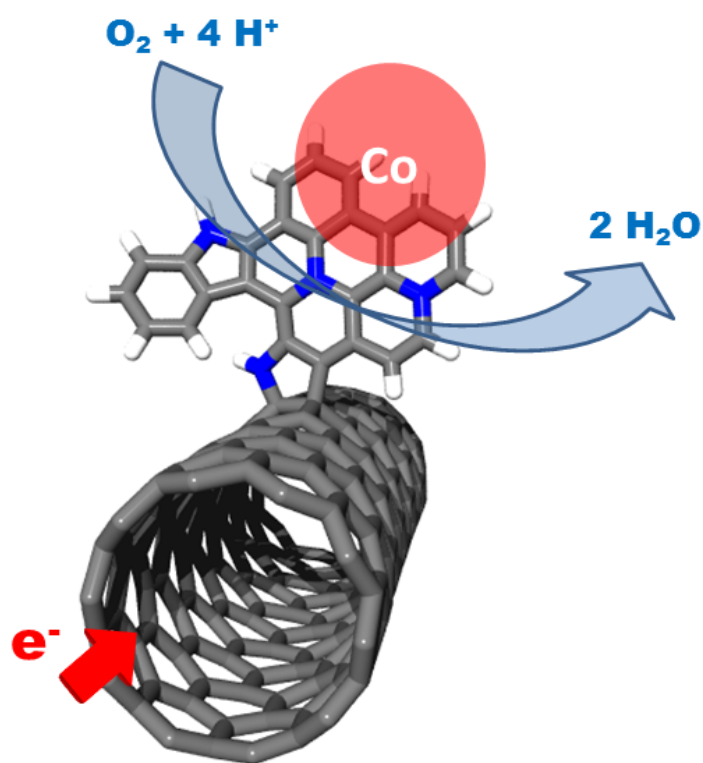
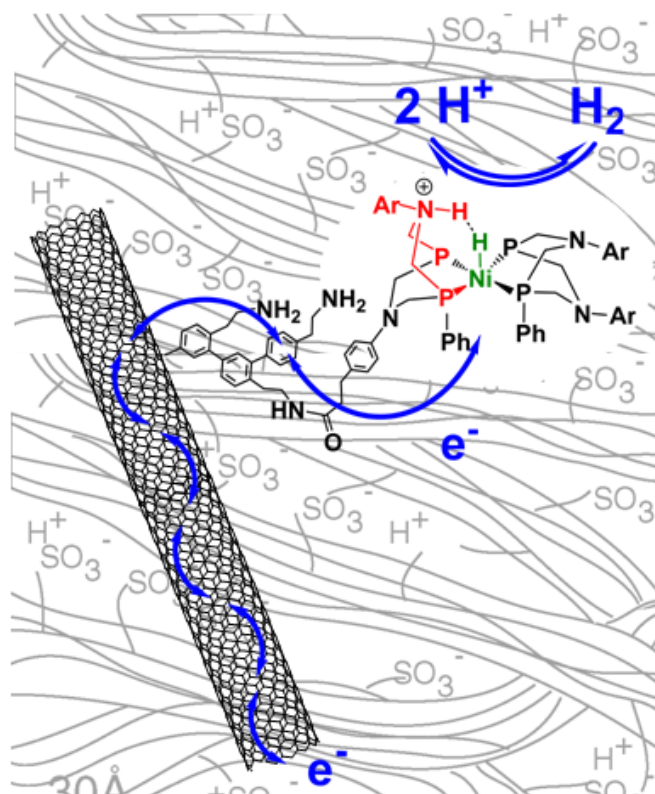
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Economically viable hydrogen uptake requests platinum-free catalysts since this expensive and scarce metal is not a sustainable resource.<sup>1</sup> A competitive alternative should be found in living micro-organisms metabolizing hydrogen thanks to specific enzymes called hydrogenases.<sup>2</sup> We will describe how the combination of a bio-inspired molecular approach with nanochemical tools,<sup>3</sup> through the attachment of mimics of hydrogenase active sites on carbon nanotubes results in nanostructured nickel-based H<sub>2</sub>-evolving cathode materials with remarkable performances: high reaction rates, bidirectional activity for H<sub>2</sub> production or oxidation without any overvoltage, exceptional stability and compatibility with the Nafion technology used for the development of PEMFC and PEM-electrolyzers.<sup>4,5</sup> We will then report on our recent studies regarding the optimization of such materials and their implementation as the anode catalyst in operational PEM fuel-cell in combination with cobalt-based cathode nanocatalysts.<sup>6-8</sup>



- (1) Chenevier, P.; Mughlerli, L.; Darbe, S.; Darchy, L.; DiManno, S.; Tran, P. D.; Valentino, F.; Iannello, M.; Volbeda, A.; Cavazza, C.; Artero, V. *C. R. Chim.* **2013**, *16*, 491-516.
- (2) Simmons, T. R.; Artero, V. *Angew. Chem. Int. Ed.* **2013**, *52*, 6143-6145.

- (3) Le Goff, A.; Moggia, F.; Debou, N.; Jegou, P.; Artero, V.; Fontecave, M.; Jusselme, B.; Palacin, S. *J. Electroanal. Chem.* **2010**, *641*, 57-63.
- (4) Le Goff, A.; Artero, V.; Jusselme, B.; Tran, P. D.; Guillet, N.; Metaye, R.; Fihri, A.; Palacin, S.; Fontecave, M. *Science* **2009**, *326*, 1384-1387.
- (5) Tran, P. D.; Le Goff, A.; Heidkamp, J.; Jusselme, B.; Guillet, N.; Palacin, S.; Dau, H.; Fontecave, M.; Artero, V. *Angew. Chem. Int. Ed.* **2011**, *50*, 1371-1374.
- (6) Morozan, A.; Jegou, P.; Jusselme, B.; Palacin, S. *Phys. Chem. Chem. Phys.* **2011**, *13*, 21600-21607.
- (7) Morozan, A.; Jegou, P.; Pinault, M.; Campidelli, S.; Jusselme, B.; Palacin, S. *Chemsuschem* **2012**, *5*, 647-651.
- (8) Morozan, A.; Jusselme, B.; Palacin, S. *Energy Environ. Sci.* **2011**, *4*, 1238-1254.